Stereochemical Studies on the Mycolic Acids

By D. E. MINNIKIN and N. POLGAR

(Dyson Perrins Laboratory, Oxford University)

In a previous communication¹ derivatives of three mycolic acids (-I, -II, and -III) from human tubercle bacilli were described and the structures (I), (II), and (III) advanced for the main components, the structure (I) corresponding to that proposed earlier² for methyl α -mycolate (Test). The esters (I) and (II) as well as the parent hydroxy-ester of (III) contain asymmetric centres (at C-2 and C-3) bearing a long alkyl chain and a hydroxyl group, respectively, and the present able, however, to detect very significant differences in the high-resolution infrared spectra (Perkin– Elmer 21 instrument, *ca.* 0.005M-solutions in carbon tetrachloride) of methyl mycolate-I and methyl 2-epimycolate-I,¹ m.p. 56–57°, $[\alpha]_{589}^{20}$ -0.6° . In the case of the epimer the ratio (0.99) of the intensity of the free hydroxyl band (3626 cm.⁻¹) to that of the bonded hydroxyl band (3534 cm.⁻¹) was greater than the corresponding ratio in the spectrum of methyl mycolate-I



studies concern their stereochemistry in respect of these centres.

Recently, Asselineau and Asselineau³ have established that corynomycolic acid has 3(D)configuration and they suggested that the centre at C-2 probably also has (D)-configuration. These authors were unable to distinguish, by their infrared spectra, α -branched β -hydroxy-esters having, respectively, *erythro*- and *threo*-configurations with respect to C-2 and C-3. We have been (0.56). Accordingly, the ratios of the free (1736 cm.⁻¹) to bonded (1719 cm.⁻¹) ester carbonyl absorptions were 1.02 and 0.39, respectively. The corresponding 1,3-diols (2-epimycolic alcohol-I and mycolic alcohol-I) gave similar results (ratios of free to bonded hydroxyl, 0.66 and 0.55, respectively). This comparison shows that the sixmembered intramolecularly hydrogen-bonded ring (presumably chair form) is destabilised in the case of methyl 2-epimycolate-I and the corresponding

1,3-diol due to an axial arrangement of one of the substituents, *i.e.*, the α -branch and the β hydroxyl group are three to one another (Fischer projection formula). Methyl mycolate-I, therefore, has erythro-configuration in respect of these centres.

The n.m.r. spectrum of the ethylidene acetal of mycolic alcohol-I showed a quartet at τ 5.38 due to the ethylidene proton, and the -CH₂-O- ring protons showed a complex pattern analogous to that in the spectrum of the ethylidene acetal of 2-hexadecylpropane-1,3-diol prepared for comparison. The spectrum of the acetal derived from the epimer showed a similar quartet at τ 5.29, indicating deshielding by the presence of an axial substituent on the same side of the six-membered ring.⁴ The pattern of the -CH₂-O- protons was disturbed, suggesting that the conformation of the ring has been destabilised by an axial These results support the assignsubstituent. ment of the erythro-configuration in respect of C-2 and C-3 in methyl mycolate-I; similar results have been obtained for mycolic esters-II and -III.

It should be noted that the above configurational assignments conflict with the views expressed earlier⁵ (see also ref. 3).

Determination of the absolute configuration at C-3 by preferential esterification of one of the antipodes of racemic α -phenylbutyric acid according to the method of Horeau⁶ indicated (D)configuration, *i.e.* the same configuration as that established3 for corynomycolic acid. Accordingly, the above mycolic esters are regarded as having 2(D), 3(D)-configurations.

(Received, August 23rd, 1966; Com. 622.)

¹.D. E. Minnikin and N. Polgar, Tetrahedron Letters, 1966, 2643.

² A. H. Etemadi and E. Lederer, Bull. Soc. chim. France, 1965, 2640.

- ³ C. Asselineau and J. Asselineau, Bull. Soc. chim. France, 1966, 1992.
 ⁴ N. Baggett, K. W. Buck, A. B. Foster, M. H. Randall, and J. M. Webber, J. Chem. Soc., 1965, 3394.
- ⁵ J. Asselineau and E. Lederer, "CIBA Foundation Symposium on Experimental Tuberculosis", Churchill, London, 1955, p. 14.

⁶ A. Horeau, Tetrahedron Letters, 1961, 506; 1962, 965; A. Horeau and H. B. Kagan, Tetrahedron, 1964, 20, 2431.